

High-coercivity ferrite magnets prepared by mechanical alloying

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Abstract

Nanocrystalline hexaferrite (BaFe₁₂O₁₉ or SrFe₁₂O₁₉) and mixed Fe,Co-ferrite ((Fe,Co_{1-x})Fe,O₄ with $x=0-1$) materials have been prepared by mechanical alloying and subsequent annealing. High coercivities were obtained in these nanocrystalline materials, 6–7 kOe for hexaferrite and \sim 3 kOe for Co-ferrite. Hexaferrite powders prepared by mechanical alloying have been used as the starting material for high-coercivity bonded magnets. Hot-pressed anisotropic hexaferrite magnets have been produced with high values of coercivity and remanence. High magnetic performance was also achieved in some mixed Fe,Co-ferrites after magnetic annealing. © 1998 Elsevier Science S.A. All rights reserved.

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tions in a wide variety of areas [1]. Ferrite-based magnetic reduction of magnetisation. In the present work, we used materials, especially $BaFe_{12}O_{19}$ and $SrFe_{12}O_{19}$, are still excess of $BaCO_3$ and $SrCO_3$ to ensure that a single the most widely used starting materials as permanent hexaferrite phase was formed. Isotropic and anisotr magnets. They have excellent chemical stability and are BaFe $_{12}O_{19}$ and SrFe $_{12}O_{19}$ magnets have been fabricated relatively cheap to produce [1,2]. Ferrite magnetic materi- by bonding and hot-pressing, respectively. The results of als with high coercivity also have applications as magnetic investigations of the dependence on composition, heat recording media [3]. treatment and magnetic annealing of the magnetic prop-

High values of coercivity may be achieved in erties of Co-ferrite are also presented here. nanocrystalline materials with a grain size close to single domain particle size [1,4]. Mechanical alloying is a unique method for the preparation of materials in nanoparticulate form [5,6]. High coercivities have been reported for **2. Experimental details** intermetallic materials based on rare-earth transition-metal compounds [6,7]. More recently, $BaFe_{12}O_{19}$ and For the synthesis of $BaFe_{12}O_{19}$ and $SrFe_{12}O_{19}$, a $SrFe_{12}O_{19}$ have been fabricated by mechanical alloying mixture of Fe_2O_3 (99% purity) and $BaCO_3$ and $SrCO_3$ [4,8,9]. These nanocrystalline materials exhibit much (99%), respectively, with compositions $1.1BaCO₃$ (or higher coercivities in comparison with hexaferrites pre-
SrCO₃)+6Fe₂O₃ were used. For Co-ferrite, Fe₃O₄ and pared using conventional powder metallurgical processes Co_3O_4 (99% for both) powders with a nominal com-
[1,2]. High magnetic performance has also been reported position of (Fe,Co_{1-x})Fe,O₄ with $x=0-1$ were mechani for Co-ferrite materials prepared by mechanical alloying cally alloyed. Mechanical alloying was carried out in a [10]. hardened steel vial together with ten 12 mm steel balls for

 $BaCO_3 + 6Fe_2O_3$ and $SrCO_3 + 6Fe_2O_3$ were used as start-
ing materials for mechanical alloying. A small quantity of $SrFe_{12}O_{19}$, all powder handling, milling and subsequent

1. Introduction hemative (Fe₂O₃) was present after formation of the hexaferrite phases, $BaFe_{12}O_{19}$ and $SrFe_{12}O_{19}$. The pres-Permanent magnet materials have found many applica- ence of the antiferromagnetic $Fe₂O₃$ phase led to a hexaferrite phase was formed. Isotropic and anisotropic

mixture of Fe₂O₃ (99% purity) and BaCO₃ and SrCO₃ position of $(Fe_xCo_{1-x})Fe₂O₄$ with $x=0-1$ were mechani-In our previous work, stoichiometric proportions of 24 h using a Spex 8000 mixer/mill. A ball to powder mass $SrFe_{12}O_{19}$, all powder handling, milling and subsequent pressing and heat treatments were performed in air. For Co-ferrite, loading of powders was performed in a pure

^{*}Corresponding author. Current address: Department of Materials argon gas filled glove box. Science, National University of Singapore, Singapore 119260. The as-milled powders were cold pressed into cylindri-

cal specimens of diameter 5 mm and length about 3 mm. The density after pressing was in the range of $3.5-4.0$ g cm⁻³ (65–75% of the theoretical density). The heat treatments, for BaFe₁₂O₁₉ and SrFe₁₂O₁₉, were carried out for 1 h at temperatures between 600 and 1050° C in air.

 $BaFe_{12}O_{19}$ and $SrFe_{12}O_{19}$ powders after annealing at 900°C were used for the starting materials for epoxy resin bonding. Epoxy resin (5 wt.%) and powder were mixed and subsequently cold pressed into cylindrical specimens with a diameter of 5 mm.

Anisotropic $BaFe_{12}O_{19}$ and $SrFe_{12}O_{19}$ samples were prepared by hot-pressing. The as-milled powder was firstly cold pressed into 5 mm diameter cylinders with a length of $4-5$ mm and a density of $3-3.5$ g cm⁻³. Cold-pressed cylinders were sealed in steel tubes and hot pressed at 900° C for 1 h, with a pressure of about 200 MPa in an Instron 1126 Uniaxial tensile tester.

 $(Fe_xCo_{1-x})Fe₂O₄$ were annealed at 750°C for 1 h. For
samples with $x \le 0.6$, the heat treatment was performed in vacuo. The samples with $x=0.7-1$ were annealed in air. After annealing the samples were maintained for 2 h at 200°C in air. Magnetic annealing was carried out at 300° C (Fe₂O₃). The non-magnetic doublet was probably due to $\frac{1}{2}$ superparamagnetism resulting from the nanocrystalline

(Siemens D5000 diffractometer with Cu K α radiation) and consisting of Fe₂O₃ and BaCO₃ [4,8,9]. The sample ⁵⁷Fe-Mössbauer spectroscopy (Canberra Packard). The samealed at 600°C consisted of hematite, and the non magnetic properties were measured at room temperature magnetic doublet is not observed (Fig. 2). The disappearusing an Oxford Instruments vibrating sample magneto- ance of the non-magnetic doublet was probably due to meter with a maximum applied field of 50 kOe. The growth of crystallite size or due to transformation of the hot-pressed cylinders were measured both parallel and amorphous phase into hematite. Samples after annealing at perpendicular to the direction of pressing. 900°C consisted of the single hexaferrite phase. It appears

3. Results and discussion

3.1. $BaFe_{12}O_{19}$ and $SrFe_{12}O_{19}$

Fig. 1 shows the X-ray diffraction patterns of $1.1BaCO₃+6Fe₂O₃$ after mechanical alloying and after subsequent annealing at 600 and 900°C, respectively. Broadened peaks associated with the hematite $Fe₂O₂$ phase were found in the X-ray diffraction patterns of the asmilled powders. The peaks for hematite became sharper after annealing at 600° C. The hexaferrite phase first appears after annealing at 800°C. Materials consisting of single hexaferrite phase were found after annealing at and above 850° C. These results are similar to those reported previously [4,8,9].

Room temperature Mössbauer spectra of as-milled and subsequently annealed $1.1BaCO₃+6Fe₂O₃$ are shown in Fig. 2. The as-milled powder has a spectrum consisting of
two components, a non-magnetic doublet and a magnetic
sextet. The magnetic sextet has a hyperfine field and other
sextet. The magnetic sextet has a hyperfine field

For Co-ferrite, the assembled powders of Fig. 1. X-ray diffraction patterns of 1.1BaCO₃+6Fe₂O₃ in the (a) as-
For Co-ferrite, the assembled powders of and after subsequent annealing at (b) 600 and (c) 900°C ((\times)

superparamagnetism resulting from the nanocrystalline The samples were examined using X-ray diffraction structure or due to the presence of an amorphous phase that the formation of the single hexaferrite phase is facilitated by having a small excess of $BaCO₃$ in the

900°C. The predicted hematite sextet and the non-magnetic doublet parameters which are identical to those of hematite referred to in the text are shown as full and dashed lines, respectively.

starting material. When stoichiometric (BaCO₃+6Fe,O₃) crystallite size. A typical crystallite size of ~300 nm was is used the product material contains a small amount of observed for samples annealed at 1000° C. High coerhematite, probably due to contamination with Fe from the civities have also been reported for nanocrystalline hexasteel vial and balls [4,8]. ferrite material prepared by other methods [1,4].

Saturation magnetisation, M_s , and coercivity, H_c , as The mechanically alloyed hexaferrite powders, annealed functions of annealing temperature, T_a , are plotted in Fig. at 900°C, were formed into isotropic bonded and hot 3. The saturation magnetisation, M_s , measured at the pressed anisotropic magnets and their properties wer 3. The saturation magnetisation, M_s , measured at the pressed anisotropic magnets and their properties were maximum field of 50 kOe was low for the as-milled measured. The isotropic bonded magnets were fabricated maximum field of 50 kOe was low for the as-milled sample and the samples resulting from annealing at from powders mixed with 5 wt.% of epoxy resin and temperatures below 600°C. This is consistent with the compacted under hydraulic pressure to form cylindrical results of X-ray and Mössbauer measurements which show samples. The hysteresis loop of the SrFe $_{12}O_{19}$ magnet is that these samples consist of antiferromagnetic hematite. shown in Fig. 4. BaFe₁₂O₁₉ and SrFe₁₂O₁₉ magnets had
The formation of the ferrimagnetic hexaferrite phase coercivities of 5.5 and 6.8 kOe, respectively. T begins at annealing temperatures around $750-800^{\circ}$ C and mum energy product was 1 MGOe which is very close to accounts for the rapid increase in saturation magnetisation the theoretically expected maximum energy product for

than the reported saturation magnetisation of 71 and 74 pared by conventional means (2–3 kOe) [1,2] which, in

emu g⁻¹ for BaFe₁₂O₁₉ and SrFe₁₂O₁₉, respectively [1,2]. some applications, may be of advantage.

1 It has been reported previously that the saturation mag-

SrFe $_{12}$ O₁₉, respectively (Fig. 3), which are similar to those reported previously for mechanically alloyed hexaferrite increasing annealing temperature, as shown in Fig. 3. tropic magnets prepared from mechanically alloyed materi-Annealing at higher temperatures led to increase of the

coercivities of 5.5 and 6.8 kOe, respectively. The maxiat these annealing temperatures (Fig. 3). isotropic hexaferrite magnets [1]. The observed coer-The saturation magnetisation values of samples annealed civities of these magnets (5–7 kOe) are significantly at 850–900°C were 65 and 68 emu g⁻¹ which are lower greater than the coercivities of hexaferrite materials pr

netisation of ultrafine particles of $BaFe_{12}O_{19}$ is smaller tionally by magnetic field alignment of the grains before than that of the bulk material, probably due to surface sintering [1,2]. The maximum energy product $(BH)_{\text{max}}$ of effects $[1,3]$. When samples are heat treated at tempera-
these magnets is \sim 4 MGOe. In the present investigation tures in the range $1000-1050^{\circ}$ C the particle size increases, anisotropic magnets were fabricated by hot pressing [1,8] the surface to volume ratio decreases and the saturation the mechanically alloyed powders described above. The magnetisation increases to a value characteristic of the densities of the magnets were greater than 95% of the bulk material.

bulk material. theoretical density of 5.3 and 5.1 g cm⁻³ for BaFe₁₂O₁₉ Samples annealed and $SrFe_{12}O_{19}$, respectively. The hysteresis loop of an anisotropic $SrFe_{12}O_{19}$ magnet is shown in Fig. 5. Coercoercivity, 5–6 kOe for BaFe₁₂O₁₉ and 6–7 kOe for anisotropic SrFe₁₂O₁₉ magnet is shown in Fig. 5. Coer-
SrFe₁₂O₁₉, respectively (Fig. 3), which are similar to those civities in the range 5.0–5.5 kOe were obse remanent magnetisations were \sim 3.6 kG, about 80% of materials [4,8,9]. The high values of coercivity are certain-
saturation magnetisation. The values of $(BH)_{\text{max}}$ for these ly associated with the microstructure. Samples annealed at anisotropic materials were in the range 2.5–3.0 MGOe 800° C had a nanocrystalline structure, with typical crys- which is significantly smaller than the values, noted above, tallite sizes of \sim 100 nm [4]. The coercivity decreases with obtained with magnetically aligned material. Again, aniso-

the annealing temperature, T_a , for Ba- and Sr-ferrite.

Fig. 3. Coercivity, *H*_c, and saturation magnetisation, σ_s , as a function of Fig. 4. Hysteresis loops of bonded Sr-hexaferrite magnet. $4\pi M_s = 4.3$ kG, the annealing temperature, T_a , for Ba- and Sr-ferrite. $4\pi M_r =$

conventional means. alloyed ferrites with compositions in the range $0.5 \le x \le 1$

may have possible applications as permanent magnets and prepared by mechanical alloying and subsequent anneal- For CoFe_2O_4 , no significant changes in remanence and ing. coercivity were observed after magnetic annealing since

structures, evidenced by line broadening observed in X-ray tropy is not possible when the B site divalent ion locations diffraction and Mössbauer studies [10]. The ordered ferrite are all occupied by Co^{2+} ions. For oth phase was obtained after annealing at 700–800°C [10]. In $x=0.5-0.8$, high remanence of 70–80% of the saturation this work, all samples were annealed at 750° C for 1 h. For $x \le 0.6$, hematite phase (Fe₂O₃) was present after annealing at 700–800°C in air. The formation of α -Fe₂O₃ was avoided by in vacuo annealing of the samples with compositions corresponding to $x \leq 0.6$. For compositions $x \geq 0.7$ in vacuo annealing gave rise to the presence of wustite (FeO) which was prevented by adopting the procedure of annealing samples, with compositions corresponding to $x \ge 0.7$, in air at 750°C.

Samples with compositions corresponding to $x \le 0.6$, after vacuum annealing at 750° C had relatively low coercivities. The coercivity was increased by further heat treatment at a lower temperature of $200-300^{\circ}$ C for 1-2 h in air. As an example it was found that the coercivity of a sample of composition ($Fe_{0.5}Co_{0.5}$)Fe₂O₄ increased from 1 to 2.2 kOe after additional heat treatment at 200° C for 2 h in air. No significant changes in structure and saturation Fig. 6. Coercivity, H_c , and saturation magnetisation, σ_s , as a function of magnetisation were found after the additional heat treat-
moposition, x, for $(F_{e_$ magnetisation were found after the additional heat treat-

ment. The increase of coercivity is probably due to increase of magnetic anisotropy energy [10–13]. Postannealing heat treatment at 200° C of samples with compositions corresponding to $x \geq 0.7$ produced no significant changes in magnetic properties.

The saturation magnetisation and coercivity, after annealing and optimum low temperature heat treatment, are plotted in Fig. 6 as functions of the composition variable, *x*. The values of saturation magnetisation of 87 and 77 emu g⁻¹ for FeFe₂O₄ and CoFe₂O₄, respectively, are close to the accepted values for the two compositions [10,11]. Samples of composition corresponding to $0 \le x \le 0.4$ have relatively low values of coercivity, being one-third of the coercivity of samples with composition parameters in the range of $0.5 \leq x \leq 1$.

The grain size of Co-ferrite samples annealed at 750° C Fig. 5. Hysteresis loops of a hot-pressed Sr-ferrite magnet in the was estimated to be \sim 25 nm from measurements of the directions parallel (———) and perpendicular (- - -) to the pressing widths of X-ray diffraction peaks and also from electron direction. 4p*M*s r c max ⁵4.4 kG, 4p*^M* ⁵3.6 kG, *^H* ⁵5.1 kOe, (*BH*) ⁵2.8 diffraction images [10]. It has been shown that the MGOe. application of a magnetic field to a mixed Fe–Co ferrite maintained at temperatures ~300°C, where significant ionic
diffusion occurs, results in directional ordering of Co^{2+} als have coercivities of 5–5.5 kOe, i.e. more than twice the ions and directional magnetic anisotropy [11] (the procoercivity of anisotropic hexaferrite magnets prepared by cedure is known as magnetic annealing). Mechanically were subjected to magnetic annealing at 300° C. The 3.2. $(Fe_xCo_{1-x})Fe₂O₄$ induced anisotropy was determined by measurement of remanent magnetisation, when the measuring field was Mixed Fe,Co-ferrites of composition $(Fe_x Co_{1-x})Fe_2 O_4$ applied in the same direction as the field for magnetic *x* have possible applications as permanent magnets and annealing, and the values of coercivity were determined recording media [2,11]. In this work, we describe the measuring fields parallel and perpendicular to the magnetic properties of materials of $(Fe_xCo_{1-x})Fe₂O₄$ with $x=0-1$ annealing field direction. The results are plotted in Fig. 7.

All the mechanically alloyed powders had disordered directional ordering of ions resulting in magnetic aniso-

Fig. 7. Coercivity, H_c , measured in the direction parallel to the direction **References** of magnetic field applied during the magnetic annealing (\parallel) and in the perpendicular direction (\perp) and remanence, σ_r , measure in the parallel direction. [1] H. Stablein, in: E.P. Wohlfarth (Ed.), Ferromagnetic Materials, Vol.

magnetisation was obtained in the direction parallel to the [2] J. Smith, H.P.J. Wijn, Ferrites, Philips' Technical Library, 1959.
[3] Z. Yang, J. Liu, P. Zheng, S. Geng, Z. Chen, IEEE Trans. Magn. 23 magnetic field applied during the magnetic annealing. (1987) 3131.

For $x=0.5-0.8$, the coercivities in the parallel direction [4] J. Ding, H. Yang, W.F. Miao, P.G. McCormick, R. Street, J. Alloys were in the range 2.8–3.2 kOe. Coercivities measured in Comp. 221 (1995) 70.
fields equino perpendicular to the direction of the field [5] C.L. Koch, Annu. Rev. Mater. Sci. 19 (1989) 2954. fields acting perpendicular to the direction of the field
applied during magnetic annealing ranged from 1.3 to 1.6 [7] J. Ding, R. Schnitzke, J. Wecker, J. Appl. Phys. 64 (1988) 5302.
(7) J. Ding, R. Street, P.G. McCormick $kOe.$ (1992) 211.

High values of remanence were obtained for samples [8] J. Ding, D. Maurice, W.F. Miao, P.G. McCormick, R. Street, J. with $x=0.5-0.8$ (Fig. 7). Using values of the theoretical Magn. Magn. Mater. 150 (1995) 417.
Agnity the movimum approve product was calculated to be [9] S.J. Campbell, W.A. Kaczmarek, G.M. Wang, Nanostruct. Mater. 6 density, the maximum energy product was calculated to be
3-4 MGOe. These values of coercivity and maximum [10] J. Ding, T. Reynolds, W.F. Miao, P.G. McCormick, R. Street, Appl. energy product are comparable with those obtained by Phys. Lett. 65 (1994) 3135. commercial Ba- and Sr-hexaferrite magnets. [11] S. Krupicka, P. Novak, in: E.P. Wohlfarth (Ed.), Ferromagnetic

Barium and strontium hexaferrites have been prepared as nanocrystals by mechanical alloying. Isotropic bonded magnets fabricated from these materials have energy products $(BH)_{\text{max}}$ ~1 MGOe and coercivities of 5.5 and 6.8 kOe for barium- and strontium-hexaferrites, respectively. Anisotropic magnets have been produced by hot pressing. Values of $(BH)_{\text{max}}$ in the range 2.5–3.0 MGOe and coercivities of 5–5.5 kOe have been achieved.

Anisotropic mixed ferrites of composition $(Fe_xCo_{1-x})Fe₂O₄$ have been prepared by magnetic annealing of nanocrystalline powders prepared by mechanical alloying. Magnets with values of remanent magnetisation of 70–80% of saturation magnetisation, coercivities above 3 kOe and $(BH)_{\text{max}}$ products in the range 3–4 MGOe have been fabricated from these powders.

- 3, North-Holland, Amsterdam, 1982.
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- Materials, Vol. 3, North-Holland, Amsterdam, 1982.
- [12] J. Ding, P.G. McCormick, R. Street, Solid State Commun. 95 (1995)
- **31. 4. Conclusions and State of the State of Table 131. (131)** V.M. Tobin, S.B. Oseroff, S. Schultz, IEEE Trans. Magn. 25 (1989) 3653.